

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

2-Chloro-4-(2-iodobenzenesulfonamido)-benzoic acid

Muhammad Nadeem Arshad,^{a*}† Islam Ullah Khan,^b H. M. Rafique,^a Abdullah M. Asiri^c and Muhammad Shafiq^b

^aX-ray Diffraction and Crystallography Laboratory, Department of Physics, School of Physical Sciences, University of the Punjab, Quaid-e-Azam Campus, Lahore 54590, Pakistan, ^bMaterials Chemistry Laboratory, Department of Chemistry, GC University, Lahore 54000, Pakistan, and ^cThe Center of Excellence for Advanced Materials Research, King Abdul Aziz University, Jeddah, PO Box 80203, Saudi Arabia
Correspondence e-mail: mnachemist@hotmail.com

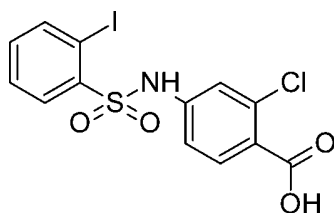
Received 28 April 2011; accepted 30 April 2011

Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.010$ Å; R factor = 0.054; wR factor = 0.129; data-to-parameter ratio = 19.2.

In the title compound, $\text{C}_{13}\text{H}_9\text{ClINO}_4\text{S}$, the dihedral angle between the aromatic rings is 81.04 (17)°. The disposition of the I and Cl atoms attached to the two rings is *anti*. In the crystal, molecules are connected via $\text{O}-\text{H}\cdots\text{O}$ and $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds.

Related literature

For background to thiazine heterocycles, see: Arshad *et al.* (2008, 2011). For their biological activity, see: Medina *et al.* (1999). For related structures, see: Arshad *et al.* (2009a,b,c).



Experimental

Crystal data

$\text{C}_{13}\text{H}_9\text{ClINO}_4\text{S}$
 $M_r = 437.62$

Monoclinic, $P2_1/n$
 $a = 14.1522$ (8) Å
 $b = 7.3203$ (4) Å
 $c = 14.7193$ (8) Å
 $\beta = 104.892$ (2)°

$V = 1473.68$ (14) Å³
 $Z = 4$

Mo $K\alpha$ radiation
 $\mu = 2.51$ mm⁻¹
 $T = 296$ K
 $0.18 \times 0.15 \times 0.09$ mm

Data collection

Bruker Kappa APEXII CCD diffractometer
Absorption correction: multi-scan (SADABS; Bruker, 2007)
 $T_{\min} = 0.661$, $T_{\max} = 0.806$

16645 measured reflections
3668 independent reflections
1876 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.058$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.054$
 $wR(F^2) = 0.129$
 $S = 1.03$
3668 reflections

191 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 1.24$ e Å⁻³
 $\Delta\rho_{\text{min}} = -1.30$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1}\cdots\text{O2}^{\text{i}}$	0.86	2.38	3.214 (6)	162
$\text{O1}-\text{H1O}\cdots\text{O2}^{\text{ii}}$	0.82	2.09	2.771 (6)	140

Symmetry codes: (i) $x - \frac{1}{2}, -y + \frac{1}{2}, z - \frac{1}{2}$; (ii) $-x + \frac{3}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$.

Data collection: APEX2 (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 (Farrugia, 1997) and PLATON (Spek, 2009); software used to prepare material for publication: WinGX (Farrugia, 1999) and PLATON.

The authors acknowledge the Higher Education Commission of Pakistan for providing a grant for the project to strengthen the Materials Chemistry Laboratory at GC University, Lahore, Pakistan.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5865).

References

- Arshad, M. N., Khan, I. U., Zia-ur-Rehman, M. & Shafiq, M. (2011). *Asian J. Chem.* **23**, 2801–2805.
Arshad, M. N., Tahir, M. N., Khan, I. U., Shafiq, M. & Sharif, H. M. A. (2009b). *Acta Cryst.* **E65**, o831.
Arshad, M. N., Tahir, M. N., Khan, I. U., Shafiq, M. & Siddiqui, W. A. (2008). *Acta Cryst.* **E64**, o2045.
Arshad, M. N., Tahir, M. N., Khan, I. U., Siddiqui, W. A. & Shafiq, M. (2009a). *Acta Cryst.* **E65**, o281.
Arshad, M. N., Tahir, M. N., Khan, I. U., Siddiqui, W. A. & Shafiq, M. (2009c). *Acta Cryst.* **E65**, o230.
Bruker (2007). SADABS, APEX2 and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
Medina, J. C., Roche, D., Shan, B., Learned, R. M., Frankmoelle, W. P., Clark, D. L., Rosen, T. & Jaen, J. C. (1999). *Bioorg. Med. Chem. Lett.* **9**, 1843–1846.
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.

† Materials Chemistry Laboratory, Department of Chemistry, GC University, Lahore 54000, Pakistan.